# Making LC Methods MS Friendly





Mark Powell

### Applications Engineer Columns and Supplies Technical Support

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# **Topics**

LC/MS ionization techniques

- •ESI
- •APCI
- •APPI

## Appropriate conditions

- Volatile buffers for MS
- •lon pair chromatography
- •HILIC

## Appropriate columns

- Column diameter
- Bonded phase
- Particle size

## Adapting existing methods to LC/MS

## Maximizing Sensitivity

- •Minimize extra column volume
- Avoiding interferences
- Sample preparation



# **LC/MS Techniques and Applications**

•Atmospheric pressure ionization (API)

•Three typical API methods:

•ESI - electrospray ionization

•APCI - atmospheric pressure chemical ionization

•APPI - atmospheric pressure photoionization

•Appropriate ionization method depends largely on analyte polarity

•Positive ion mode (protonation) or negative ion mode (deprotonation)

•Masses measured as mass to charge ratio (m/z)



# **Applicability of Atmospheric Pressure Ionization Techniques**



API-ESI = Atmospheric pressure electrospray ionization APCI = Atmospheric presure chemical ionization APPI = Atmospheric pressure photo ionization



# **Electrospray Ionization**

- Most common ionization technique
- Used for high and low molecular weight compounds
- Ions are formed in solution and then the droplets are evaporated
- Analyte volatility not required
- Compounds containing heteroatoms such as N, S, and O typically analyze well
- Can form multiply-charged ions
- Like UV detection, ESI is concentration sensitive
- ESI is generally more sensitive for samples that are ionized in solution



# **Electrospray Ionization**





# **APCI and APPI Sources**

•APCI

•Analyte and mobile phase are first evaporated then ionized by corona needle

•Good technique for low to medium polarity analytes

- •High probe temperatures desolvate and vaporize the sample
- •Could lead to sample decomposition
- •Not a good choice for thermally unstable analytes
- •Forms singly charged ions

•APPI

- •Analyte and mobile phase are first evaporated then ionized with light
- Good technique for hydrophobic conjugated ring systems
- thermally sensitive compounds
- •May be less susceptible to ion suppression than ESI



# **APCI and APPI Sources for the LC/MSD**





# **Multimode Source**



Capable of simultaneously generating ions by electrospray and APCI
Positive ESI, negative ESI, positive APCI, and negative APCI in a single run



# **Multimode Source**





# **Method Considerations for ESI**

- pH of the mobile phase (and analyte pKa) affects ion formation
- Voltage applied to the electrospray probe will induce ion formation
- Choosing best mobile phase pH analytes can improve sensitivity
- Organic solvent has little effect on ionization
- Works best with buffer concentrations below 25 mM
- Works best at low flow rates (less than 0.5 mL/min)
  - 5 µL/min up to 2 mL/min (for ESI with Agilent Jet Stream thermal gradient focusing)
- Compatible with reversed phase, HILIC, normal phase



# **Buffer Considerations for ESI**

•Buffer concentrations below 25 mM (best below 10 mM)

•Poor compatibility with non-volatile buffers

•Deposit buildup

•Metal ion buffers interfere with ionization

•Acidic mobile phases generally favor positive mode ionization

•0.1% - 1% formic acid, 0.1% - 1% acetic acid, 0.05% - 0.2% TFA

•Ammonium salts (ammonium formate and ammonium acetate) favor formation of ammonium adducts

•TFA causes ion suppression

•Use TFA "fix" – post column addition of acetic or propionic acid

•Basic mobile phases generally favor negative mode ionization

•Ammonium hydroxide, triethylamine, diethylamine, piperidine, ammonium bicarbonate

•pH 1 to 2 units away from the pKa of the analytes



# Method Considerations for APCI and APPI

- •LC mobile phase solvents can interfere with ionization
- •Try methanol first (acetonitrile can be a problem)
- •Poor compatibility with non-volatile buffers
- •Works with wider range of buffer concentrations than ESI
- •Less than 100 mM
- •Broader range of flow rates, up to 1.5 mL/min
- •Higher sensitivity and less noise than ESI at flow rates >0.75 mL/min
- •Highly flammable solvents should be avoided



## **Effects of Volatile and Non-Volatile Buffers**







Theophylline (TP) M.W=180.17 pKa <1, 8.6

Theobromine (TB) M.W=180.17 pKa <1, 10.0

Caffeine (CF) M.W=194.19 pKa = 14



# **Volatile Buffer vs. Non Volatile Buffer**



Column: ZORBAX Eclipse XDB-C18 2.1 x 150 mm, 5µm Mobile Phase: 1) 5mM AcONH4 (pH 4.6)/MeOH=80:20 2) 5mM KH2PO4 (pH	LC condition	<u>IS</u>	
2.1 x 150 mm, 5µm Mobile Phase: 1) 5mM AcONH4 (pH 4.6)/MeOH=80:20 2) 5mM KH2PO4 (pH	Column:	ZORBAX Eclipse XDB-C18	
Mobile Phase: 1) 5mM AcONH <sub>4</sub> (pH 4.6)/MeOH=80:20 2) 5mM KH <sub>2</sub> PO <sub>4</sub> (pH		2.1 x 150 mm, 5µm	
4.6)/MeOH=80:20 2) 5mM KH2PO4 (pH	Mobile Phas	e: 1) 5mM AcONH4 (pH	
2) 5mM KH2PO4 (pH	4.6)/MeOH=80:20		
	-	2) 5mM KH2PO4 (pH	
2.5)/MeOH=80:20	2.5)/MeOH=	80:20	
Flow rate: 0.2mL/min	Flow rate:	0.2mL/min	
Temp: 40°C	Temp:	40°C	
Inj.volume: 5µL	Inj.volume:	5µL	

MS conditionsIonization:ESIMode:PositiveMass range:m/z 100~200Capillary volt.:3.5kVFragmentor volt :100VDrying gas:N2 (12.0L/min ,350°C)Nebulizer gas:N2(50psi)N2



# **APCI Signal After 600 Injections of Salt Solution**



• Initial instability in the signal is probably due to changing electric fields as salt deposits in the source.



# The Effects of Having Non-volatile Buffers in the Mobile Phase





Cleaning the spray chamber



# **Effect of Volatile Buffer Concentration on ESI**

Lower buffer concentrations provide better droplet evaporation





# **Effect of Volatile Buffer Concentration on APCI**



APCI



SIM: 195.2 and 609.3

Vcap: 4000V

Vaporizer: 400C

Nebulizer: APCI - 60 psig

Drying gas: APCI - 350 C, 5 L/min

Fragmentor: Ramped 70 V for 195.2; 120 V for 609.3

# pH Effects on Selectivity and MS Sensitivity





# Ion Pair Chromatography and LC/MS

- •Mobile phase includes an ion-pair reagent
- •Hydrophobic portion adsorbs to stationary phase
- Ionic portion pairs with the analyte
- •Alkyl sulfonates or tetraalkyl ammonium salts
- •Non-volatile
- •Ion pair reagents can interfere with ionization process
- •Use heptaflurobutyric acid (HFBA) and tributylamine (TBA)
- •HILIC can be an alternative



# HILIC vs. Reversed Phase – ESI sensitivity



Agilent 1290 Infinity LC System Agilent 6410A LC/MS A: 10 mM ammonium formate pH 3.2 B. acetonitrile / 100 mM ammonium formate pH 3.2 (9:1) 0.4 mL/min Isocratic elution Injection Volume: 2 µL Column: 25 °C MS: ESI+, SIM, 250 °C, 11 L/min, 30 psi, 4000 V, 200 V delta EMV. 20 ms dwell time Sample: Normorphine, m/z 272 Morphine, m/z 286 Morphine-6-B-D-glucuronide (M6G), m/z 462 Morphine-3-B-D-glucuronide (M3G), m/z 462

5991-1242EN



# **Agilent LC-MS Column Configurations**



Column Type	Column I.D.	Typical Flow Rate Range
Analytical	4.6 mm	1 – 1.5 mL/min
Solvent Saver	3.0 mm	0.3 – 1 mL/min
NarrowBore	2.1 mm	0.1 – 0.5 mL/min
MicroBore	1.0 mm	0.03 – 0.2 mL/min
Capillary	0.3, 0.5 mm	2 – 40 µL/min
Nano	0.075, 0.10 mm	0.1 – 0.6 µL/min



# **Column Choices for LC/MS Analysis**



	TFA	Formate/ Formic Acid	Acetate/ Acetic Acid	Ammonium Hydroxide
Eclipse Plus	$\checkmark$	$\checkmark$	$\checkmark$	X
StableBond	$\checkmark$	$\checkmark$	$\checkmark$	X
Eclipse XDB	$\checkmark$	$\checkmark$	$\checkmark$	X
Bonus-RP	$\checkmark$	$\checkmark$	$\checkmark$	X
Extend-C18	✓	$\checkmark$	$\checkmark$	$\checkmark$
HILIC Plus	$\checkmark$	$\checkmark$	$\checkmark$	X



# **Poroshell 120 Phases**



# Superficially porous microparticulate column packing

Poroshell 120 particles have a 1.7  $\mu$ m solid silica core with a 0.5  $\mu$ m porous outer layer to make a 2.7  $\mu$ m particle. This carefully selected configuration gives you all the performance advantages of sub-2  $\mu$ m particles with backpressure that is comparable to a sub-3  $\mu$ m particle.

The Measure of Confidence



•SB-Aq •Bonus-RP •Phenyl-Hexyl •HILIC





#### 5990-8795EN



# Particle size and LC/MS performance



Agilent 1290 Infinity LC System Agilent 6410A LC/MS A: 10 mM ammonium formate pH 3.2 B: acetonitrile / 100 mM ammonium formate pH 3.2 (9:1) 0.4 mL/min Isocratic elution, 10% B Injection Volume: 2 µL Column: 25 °C MS: ESI+, SIM, 250 °C, 11 L/min, 30 psi, 4000 V, 200 V delta EMV, 20 ms dwell time Sample: Morphine-6- $\beta$ -D-glucuronide (M6G), m/z 462 Morphine-3- $\beta$ -D-glucuronide (M3G), m/z 462



# Particle Size and LC/MS performance





# Particle Size and LC/MS performance











1	SMR	6	SMMX
2	PYM	7	DFZ
3	TCP	8	SDMX
4	SDD	9	SOX
5	FZD	10	OXA

Instrument:	Agilent 1100 Series HPLC		
Column:	250 mm × 4 mm id, RP-18 Purospher, 5 μm, p/n 79925PU-584		
Mobile phase:	$A = 0.7\%$ Phosphoric acid, $B = CH_3CN$		
Gradient:	0.0 min 5% B; 10.0 min 5% B; 40.0 min 65% B; 45.0 min 65% B; Post Time 7.0 min 5% B		
Flow rate:	1.0 mL/min		
Temperature:	40 °C		
Injection volume:	20 µL		

5988-7135EN





5990-6238EN





Agilent Poroshell 120 EC-C18 4.6 mm × 50 mm, 2.7 µm

nm, 8 nm, ref off 3 mm, 2 uL micro flow cell; Peak width >0.05 min. (40Hz)

5990-6238EN





•Conditions were scaled for a 3.0 x 50 mm column

•Shows that 3.0 mm can easily be used for conventional UV and MS detection

5990-6238EN



# **Analysis of Ten Compounds Found in Green Tea**



5990-7824EN



# **Analysis of Ten Compounds Found in Green Tea**



A = 0.1% H3PO4 in H2O B = CH3CN 1 mL/min 40 °C Sig = 210,4 nm, Ref = Off 2- $\mu$ L, 3-mm micro flow cell Sample: 0.03 mg/mL each in H2O/CH3CN **4.6 × 150 mm Zorbax SB-C18, 5 \mum** 0.0 min, 10% B; 7.5 min, 15% B; 15 min, 27% B 15  $\mu$ L injection

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# **Optimizing Sensitivity with Mobile Phase Selection**



Agilent 1200 Series RRLC / 6410A Triple Quadrupole MS; Poroshell 120 SB-C18, 2.1 x 100 mm; Acidified Water / Acetonitrile Gradient, 0.7 mL/min; ESI-, SIM, 350 °C, 10 L/min, 50 psi, -3500 V



# **Extra Column Volume and Sensitivity**





#### Agilent 1290 Infinity Binary LC system containing 6140 Single Quad MS

Column: Agilent ZORBAX SB C18, 50 × 2.1 mm, 1.8  $\mu$ m Solvent A: Water + 0.1% formic acid; Solvent B: Acetonitrile + 0.1% formic acid; Flow rate: 0.5 mL/min Gradient : 0 min 10% B; 5 min 20% B; 5.01 min 95% B; 6 min 95% B Injection volume: 1  $\mu$ L; Column temperature: 40 °C; Source: Gas temperature: 350 °C, nebulizer pressure: 45 psi, gas flow: 11 L/min, positive polarity, Scan: 100 – 1000 *m/z* Sample: Solution of Sulfamethizole (first peak, *m/z 271.0), Sulfamethazine* (second peak, *m/z 279.0), Sulfachloropyridazine (third peak, m/z 285.0), Sulfadimethoxine (fourth peak, m/z 311.0) each at a* concentration of 100 ng/ $\mu$ L.









# **Extra Column Volume and Sensitivity**





# **Extra Column Volume and Sensitivity**





# **Avoiding Interferences and Ion Suppression**



Agilent 1200 Series RRLC / 6460A Triple Quadrupole MS; Ammonium Acetate pH 5 / Acetonitrile Gradient, 0.4 mL/min; ESI+, Scan 100-800, 400 °C, 12 L/min, 40 psi, 3500 V;

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# **Interferences and Sample Prep**





# **Interferences and Sample Prep**







# **Preventing Instrument Down Time**



Clean Source

After SPE

After PPT



# Summary

•ESI is most common atmospheric pressure ionization technique

- •APCI and APPI for less polar molecules that do not ionize well by ESI
- •Flow rate and mobile phase buffer selection are important for best LC/MS performance
- •Importance of choosing column ID and phase for the best results
- •Column particle size and efficiency can avoid interferences and maximize sensitivity
- •Removing interferences when avoiding them is not enough

Contact Tech Support 1-800-227-9770 Ic-column-support@agilent.com

