



Liquid Chromatograph-Mass Spectrometer LCMS[™]-9030

Analysis of Impurities in Pharmaceuticals Using LCMS-9030 Quadrupole Time-of-Flight Liquid Chromatograph-Mass Spectrometer

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User Benefits

- Highly accurate qualitative analysis for impurities in pharmaceuticals is possible by using this system.
- ◆ The analysis software LabSolutions Insight Explore[™] can be used for composition estimation, database search for the structural formula, and fragment assignment analysis.

Introduction

It is extremely important to identify trace impurities in pharmaceuticals to guarantee their quality and safety. The structural formulas of impurities in pharmaceuticals are described in official documents such as the Japanese Pharmacopoeia (JP), the European Pharmacopoeia, and the United States Pharmacopeia. In general, the HPLC-UV method is widely used for analysis of impurities. In contrast, structural analysis using a mass spectrometer such as LC/MS/MS is attracting attention as a useful means for identifying detected impurities.

This application news introduces an example of structural analysis of the impurities in montelukast sodium using the quadrupole time-of-flight (QTOF) mass spectrometer LCMS-9030 (Fig. 1) and the software LabSolutions Insight Explore. Montelukast sodium is listed in the 17th edition of the JP and is used as a therapeutic drug for bronchial asthma and allergic rhinitis.

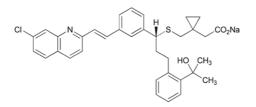


Fig. 1 Exterior of Nexera[™] X3 and LCMS[™]-9030

Analysis of Montelukast Sodium

According to the preparation procedure for montelukast sodium (Fig. 2) described in the Japanese Pharmacopoeia (JP), solution A (1 mg/mL) was prepared using the montelukast standard for system suitability test. Table 1 lists the analytical conditions.

Fig. 3 shows the obtained UV chromatogram. Montelukast, the principal component, was eluted at the retention time of approximately 10 min, and impurity peaks (1) to (5) were detected before and after that.



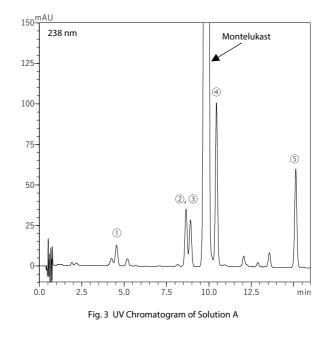


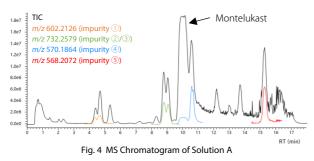
Table 1 Analytical Conditions

[UHPLC conditions (N	levera X3)]				
Column : Shim-pack Scepter TM Phenyl-120,					
Column					
	50 mm×2.1 mm l.D., 1.9 μm ^{*1}				
Mobile phase A	: Water/Formic acid = 2000/3				
Mobile phase B	: Acetonitrile/Formic acid = 2000/3				
Flow rate	: 0.25 mL/min				
Time program (%B)	: 45% (0–3 min) → 65% (16 min) →				
	45% (16.1–25 min)				
Column temp.	: 30 °C				
Injection volume	: 10 μL				
Detection	: UV 238 nm				
[MS conditions (LCM	5 conditions (LCMS-9030)]				
lonization	: ESI positive				
Mode	: MS, MS/MS scan				
Nebulizing gas flow	: 3.0 L/min				
Drying gas flow	: 10.0 L/min				
Heating gas flow	: 10.0 L/min				
DL temp.	: 250 °C				
BH temp.	: 400 °C				
Interface temp.	: 300 °C				

*1: P/N 227-31063-03

Composition Estimation for Impurities

Peaks for impurities, corresponding respectively to the retention time detected by UV, were observed on the extracted ion chromatogram (XIC) at m/z 602.2126, 732.2579, 570.1864 and 568.2072 by mass spectrometry. Fig. 4 shows the total ion chromatogram (TIC) of solution A and the XIC of each impurity.



Based on the obtained MS spectra, composition estimation was performed using Insight Explore. As an example, Fig. 5 shows the composition estimation results for the impurity peak 4 (m/z 570.1864). It was found that the compositional formula with the highest score is C₃₄H₃₂NO₃SCI.

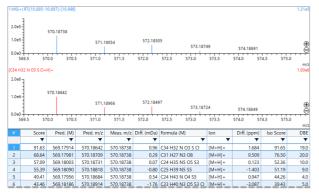


Fig. 5 Results of Composition Estimation for Impurity 4(Top: Observed MS Spectrum, Middle: Theoretical Spectrum, Bottom: Candidates of Compositional Formula)

Compound Search and Fragment Assignment

An analysis was carried out using the "Assign" function of Insight Explore to confirm the structural formula and the compound name of impurity peak ④. This software makes it possible to list compounds from the online ChemSpider™ database based on compositional formula information. Then, by conducting the assignment for the obtained compounds, the degree of coincidence (assign score) between the product ions obtained via fragment prediction and those observed in the measured MS/MS spectrum is calculated.

An online database search for the compositional formula C₃₄H₃₂NO₃SCl found six candidate compounds (Fig. 6). The topranked compound was Montelukast methyl ketone (ChemSpider ID 17623689). This compound was found to be same as the impurity E described in the JP: [1-({[(1R)-3-(2-Acetylphenyl)-1-{3-[(E)-2-(7-chloro-2-quinolinyl)vinyl]

phenyl]propyl]sulfanyl}methyl)cyclopropyl]acetic acid.

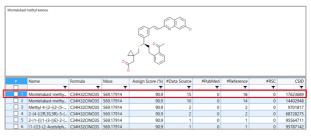


Fig. 6 Results of ChemSpider Database Search for C₃₄H₃₂NO₃SCI

Next, Fig. 7 shows an example of automatic assignment for fragment ions by the "Assign" function.

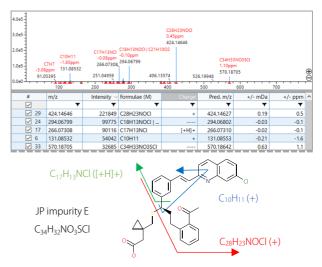


Fig. 7 Results of Fragment Assignment for Impurity E

Finally, the LCMS-9030 analytical results for each impurity are summarized in Table 2. It was also found that each of the impurity peaks other than 4 corresponds to the impurities described in the JP. Moreover, it was possible to analyze the impurities with high mass accuracy within 1 mDa of mass error compared to the theoretical mass value.

Impurity peak	HPLC RT (min)	JP listed impurities	[M+H] ⁺ theoretical	[M+H] ⁺ observed	Error (mDa)
1	4.55	А	602.2126	602.2132	0.59
2/3	8.65/8.92	C/D	732.2579	732.2584	0.52
-	9.97	Montelukast	586.2177	586.2179	0.15
(4)	10.45	E	570.1864	570.1874	0.96
5	15.13	F	568.2072	568.2079	0.70

■ Conclusion

By using a Shimadzu LCMS-9030 quadrupole time-of-flight liquid chromatograph-mass spectrometer and an analysis software LabSolutions Insight Explore, it is possible to analyze the structure of impurities contained in pharmaceuticals. This procedure is expected to contribute to the analysis of trace impurities in products in other fields such like food and chemical industries.

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