

# Revisions to EPA Method 624 for Analysis of VOCs by GC/MS

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# Introduction

Environmental contamination has been at the forefront of government policy and regulation since the US EPA was established in 1970. Over the years the US EPA has developed, published, and updated multiple methods for analysis of environmental pollutants, and single-quadrupole gas chromatography-mass spectrometry (GCMS) has long been the technique of choice for determination of volatile organic contaminants (VOCs). The GCMS instrumentation has evolved, with improvements in sensitivity, reliability, and user experience, but there haven't been any significant

advancements in the overall methodology since the mid-1980s.

The US EPA is currently in the process of developing a revision to US EPA Method 624, which was first promulgated in 1984 and specified the use of packed columns as part of the protocol to collect data on the VOC pollutants. This poster describes a GCMS purge-and-trap (P&T) method validation study conducted to evaluate operating conditions for the existing US EPA Method 624 VOC list, using updated technology and advanced GCMS instrumentation.



Figure 1: Shimadzu GCMS-QP2010 SE

# Experimental

This study was conducted using the Shimadzu GCMS-QP2010 SE shown in Figure 1, configured with a capillary column designed specifically for analysis of VOCs by US EPA Method 624. The GC was operated in the unique Constant Linear Velocity mode to provide

optimum chromatographic resolution, symmetric peak shape, and enhanced sensitivity for all compounds. Changes to quantitation and reference ions for a few selected compounds were made to improve overall sensitivity of the method.

# Results and Discussion

## BFB Tune Results

At the beginning of the project the GCMS-QP2010 SE was tuned to meet the US EPA Method 624 requirements. Each day prior to running samples, and at intervals of no longer than 12-hours during long sequences, an aliquot of the 4-bromofluorobenzene (BFB)

was purged and analyzed. The BFB spectra were evaluated using the US EPA Method 624 criteria. A representative example of a BFB chromatogram and spectrum are shown in Figure 2.

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Table 1 lists the BFB results as compared to the method criteria from three selected analyses of BFB during one extended sequence. The tune remained stable for over

2½ months, and the GCMS-QP2010 SE instrument did not require re-tuning at any time during the analysis period.

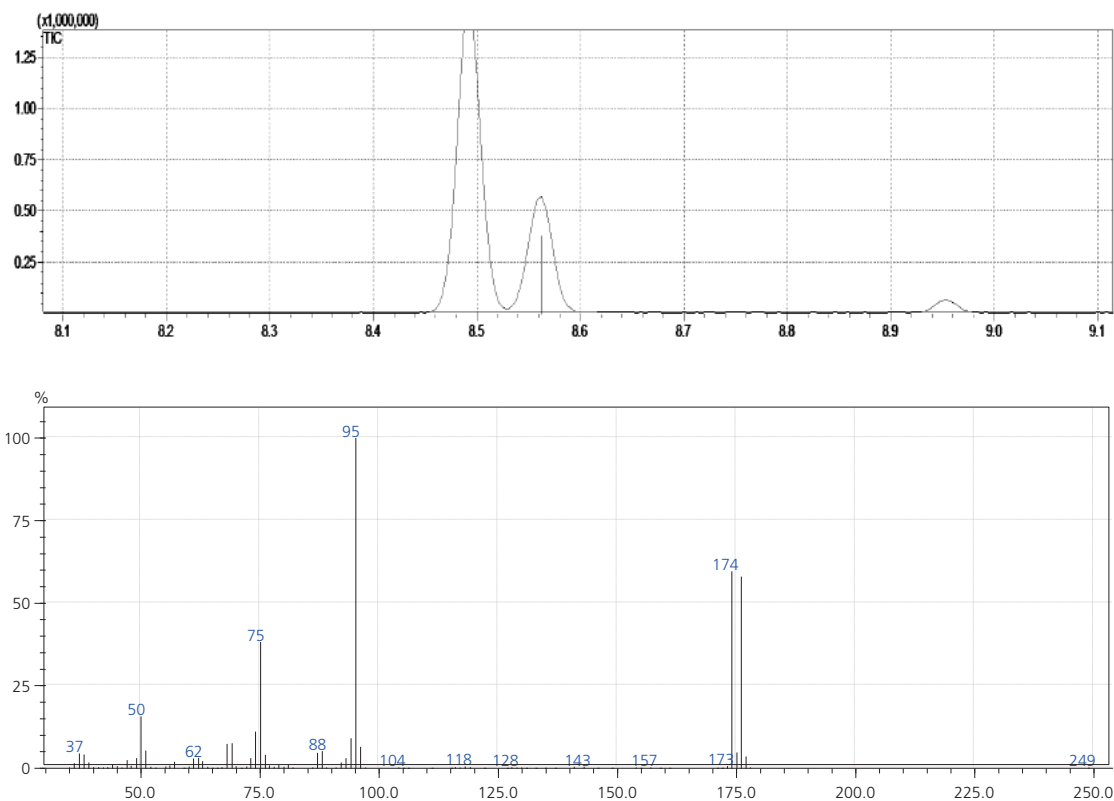


Figure 2: Typical Chromatogram and Spectrum from BFB Tune Evaluation

Table 1: Evaluation of BFB Spectra from 3 Different Runs across a Long Sequence, Compared to US EPA Method 624 Criteria

m/z	Spectrum Check Criteria	Result		Result		Result	
		Run #1	Status	Run #10	Status	Run #30	Status
50	15 to 40% of mass 95	16.0	Pass	15.6	Pass	15.4	Pass
75	30 to 60% of mass 95	41.2	Pass	41.5	Pass	43.5	Pass
95	Base Peak, 100% Relative Abundance	100.0	Pass	100.0	Pass	100.0	Pass
96	5 to 9% of mass 95	6.5	Pass	6.9	Pass	7.0	Pass
173	< 2% of mass 174	1.0	Pass	1.6	Pass	1.5	Pass
174	> 50% of mass 95	60.8	Pass	59.3	Pass	61.6	Pass
175	5 to 9% of mass 174	7.5	Pass	7.4	Pass	7.4	Pass
176	> 95% but < 101% of mass 174	97.2	Pass	100.6	Pass	97.0	Pass
177	5 to 9% of mass 176	6.6	Pass	5.6	Pass	6.4	Pass

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### Initial Demonstration of Proficiency

A series of nine calibration standards across the range of 0.5 to 200 µg/L (parts-per-billion, ppb) was prepared. The three internal standards (IS) were held constant at 30 µg/L, and the three surrogate standards (Surr) were held constant at 10 µg/L in all samples analyzed. A total ion

chromatogram (TIC) from the 10 µg/L standard is shown in Figure 3, along with an expanded view of the chromatography of the early-eluting light gases. Results from the initial demonstration of proficiency are shown in Table 2.

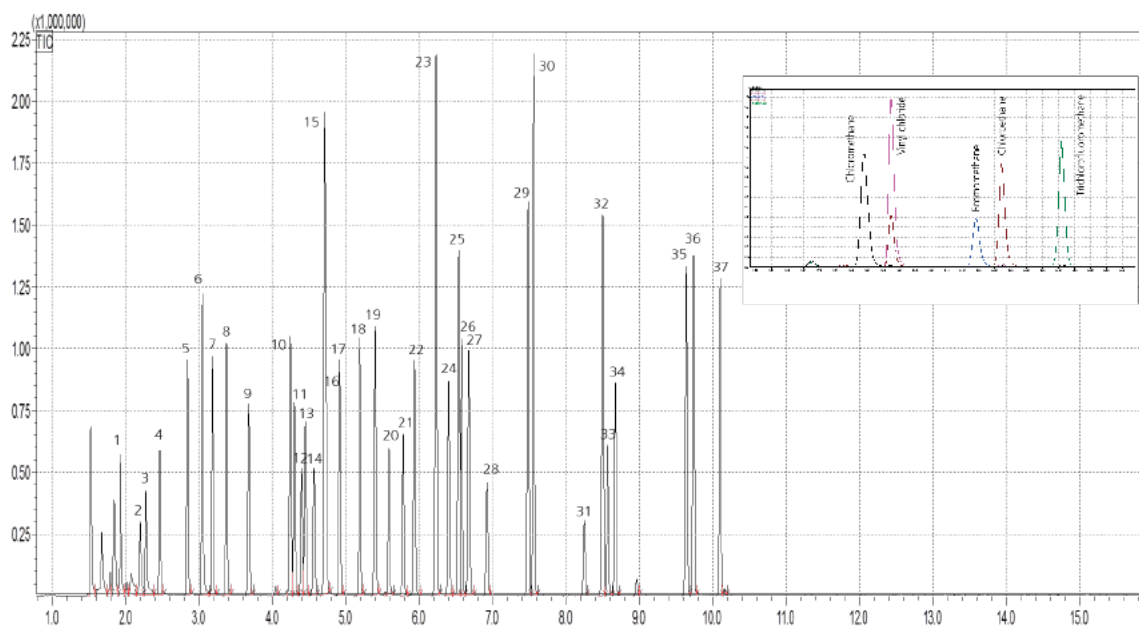


Figure 3: Total Ion Chromatogram from the 10 µg/L Calibration Standard and EICP of the Five Light Gases. Peak numbers correspond to compound names shown in Table 2.

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Table 2: Statistical Results from the Initial Demonstration of Proficiency

Peak #	Compound Name	9-Point Calibration			Continuing Calibration			Method Detection Limit		Precision and Accuracy		
		0.5 to 200 µg/L			Calculated Concentration			0.5 µg/L n = 8		0.5 µg/L n = 8		
		R <sup>2</sup>	Avg RF	RF % RSD	CCV#1 µg/L	CCV#2 µg/L	CCV#3 µg/L	% RSD	MDL	Mean Concentration (µg/L)	Recovery	%RSD
1	Chloromethane	0.9995	3.03	9.0	8.0	8.1	8.2	8.6	0.14	0.55	110%	8.6
2	Vinyl chloride	0.9998	3.23	7.7	8.4	8.4	8.5	5.5	0.08	0.52	103%	5.5
3	Bromomethane	0.9998	1.25	9.6	8.3	8.1	8.3	11.5	0.21	0.61	122%	11.5
4	Chloroethane	0.9999	2.16	8.6	8.7	8.7	8.6	4.8	0.08	0.53	106%	4.8
5	Trichlorofluoromethane	0.9998	2.64	7.9	8.7	8.8	8.7	8.2	0.14	0.56	112%	8.2
6	1,1-Dichloroethene	1.0000	2.11	6.1	9.1	9.1	9.2	5.4	0.08	0.51	102%	5.4
7	Methylene chloride	1.0000	3.36	5.8	9.2	9.2	9.1	6.2	0.10	0.55	110%	6.2
8	trans-1,2-Dichloroethene	0.9999	2.35	6.8	9.0	9.0	9.0	8.0	0.12	0.52	104%	8.0
9	1,1-Dichloroethane	1.0000	4.43	4.2	9.3	9.3	9.4	4.8	0.07	0.50	100%	4.8
10	Bromochloromethane (IS)	NA	NA	NA	NA	NA	NA	NA	NA	30.00	NA	NA
11	Chloroform	1.0000	3.45	4.5	9.3	9.3	9.3	4.6	0.07	0.52	104%	4.6
12	Pentafluorobenzene (Surr)	NA	3.85	3.2	9.7	9.6	9.7	1.1	0.34	9.86	99%	1.1
13	1,1,1-Trichloroethane	0.9999	2.56	3.0	9.4	9.4	9.3	8.1	0.11	0.46	91%	8.1
14	Carbon tetrachloride	0.9991	1.80	7.6	9.2	9.2	9.3	4.6	0.06	0.41	81%	4.6
15	Benzene	0.9993	10.20	4.5	9.3	9.3	9.3	2.8	0.04	0.50	99%	2.8
16	1,2-Dichloroethane	1.0000	2.83	6.8	9.2	9.3	9.2	3.8	0.06	0.52	105%	3.8
17	Fluorobenzene (Surr)	NA	10.54	2.4	9.8	9.8	9.8	1.4	0.40	9.77	98%	1.4
18	Trichloroethene	1.0000	2.24	2.8	10.4	10.4	10.4	3.3	0.05	0.52	104%	3.3
19	1,2-Dichloropropane	0.9997	0.74	3.9	9.5	9.4	9.5	3.1	0.05	0.53	105%	3.1
20	Bromodichloromethane	0.9997	0.70	4.2	9.4	9.3	9.4	7.1	0.10	0.49	98%	7.1
21	2-Chloroethylvinyl ether	0.9999	0.13	10.9	9.7	9.6	9.8	12.4	0.17	0.47	93%	12.4
22	cis-1,3-Dichloropropene	0.9999	0.92	8.7	9.0	8.9	9.2	4.8	0.07	0.46	92%	4.8
23	Toluene	0.9996	1.76	6.9	9.7	9.7	9.7	5.7	0.08	0.47	93%	5.7
24	trans-1,3-Dichloropropene	0.9998	0.80	11.1	8.8	8.6	9.0	7.6	0.10	0.45	90%	7.6
25	2-Bromo-1-chloropropane (IS)	NA	NA	NA	NA	NA	NA	NA	NA	30.00	NA	NA
26	1,1,2-Trichloroethane	0.9999	0.63	6.4	9.3	9.4	9.3	3.6	0.06	0.54	108%	3.6
27	Tetrachloroethene	0.9976	0.43	8.8	9.3	9.5	9.2	4.6	0.07	0.52	104%	4.6
28	Dibromochloromethane	0.9997	0.36	6.7	9.4	9.3	9.5	5.4	0.08	0.49	98%	5.4
29	Chlorobenzene	0.9996	1.38	2.5	9.7	9.7	9.7	4.7	0.07	0.50	100%	4.7
30	Ethylbenzene	0.9978	0.80	8.0	10.3	10.4	10.4	3.6	0.04	0.41	83%	3.6
31	Bromoform	0.9997	0.23	10.4	9.7	9.5	9.9	7.5	0.10	0.46	93%	7.5
32	1,4-Dichlorobutane (IS)	NA	NA	NA	NA	NA	NA	NA	NA	30.00	NA	NA
33	4-Bromofluorobenzene (Surr)	NA	0.81	2.2	10.2	10.2	10.2	0.7	0.21	10.15	101%	0.7
34	1,1,1,2-Tetrachloroethane	1.0000	0.80	3.6	8.3	8.3	8.2	5.4	0.08	0.51	103%	5.4
35	1,3-Dichlorobenzene	0.9979	0.93	5.2	10.0	10.1	9.9	4.8	0.07	0.51	102%	4.8
36	1,4-Dichlorobenzene	0.9975	0.92	5.4	10.2	10.3	10.2	5.1	0.08	0.51	101%	5.1
37	1,2-Dichlorobenzene	0.9979	0.87	5.4	10.3	10.4	10.3	5.6	0.08	0.50	101%	5.6

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### Stability of Internal and Surrogate Standards

Internal standard response remained stable during the entire study at  $\leq 4\%$ , and Surrogate recoveries fell within the 80 – 120 % method criteria for all analyses. IS and

Surr results from a representative 12-hour sequence are shown in Figures 4 and 5, respectively.

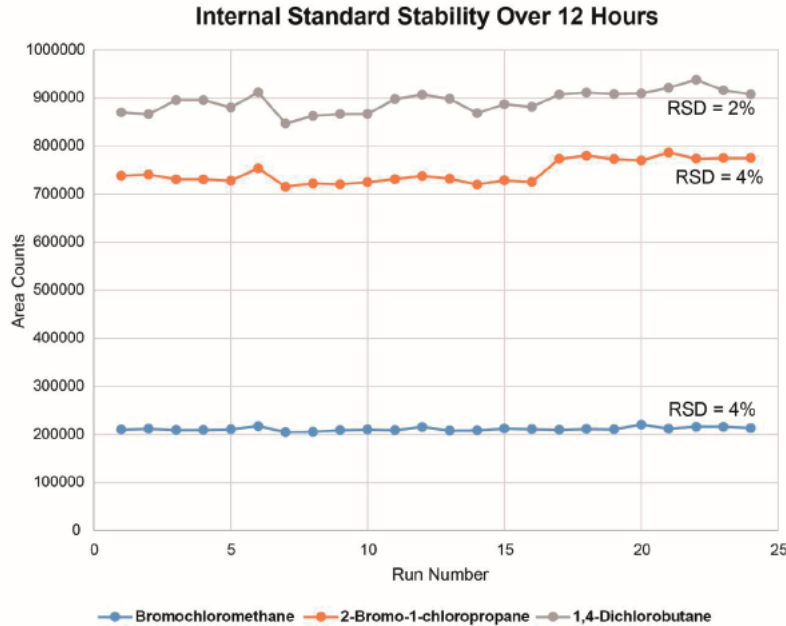


Figure 4: Internal Standard Response of a Representative 12-Hour Period during This Study

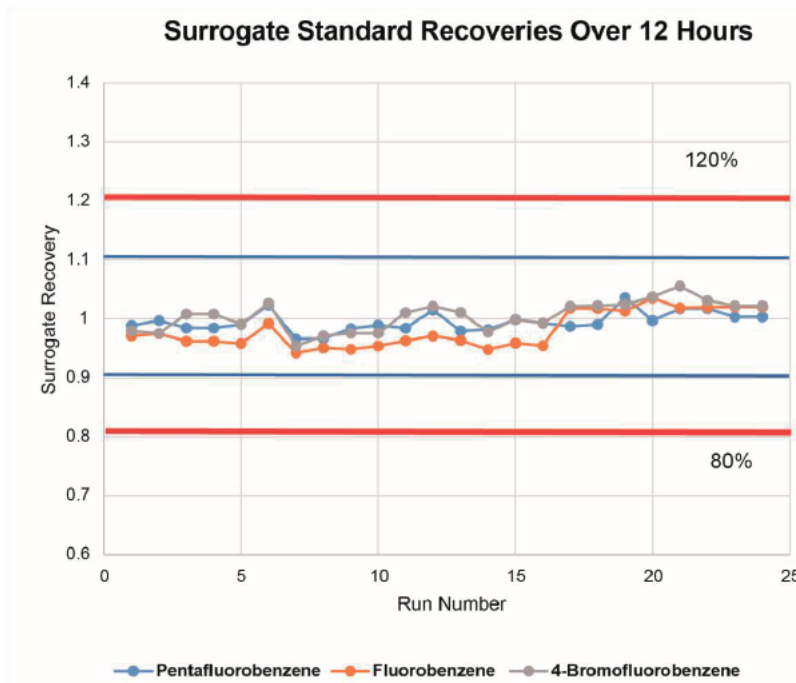


Figure 5: Surrogate Standard Recoveries over a Representative 12-Hour Tune Period during This Study

## Summary and Conclusions

The instrumentation and analytical conditions shown here have been demonstrated to provide outstanding results for US EPA Method 624, far exceeding all existing method criteria. The narrow-bore capillary column and Constant Linear Velocity mode provided outstanding chromatography for all compounds, including the early-eluting light gases, in less than 10 minutes.

Calibration curves over narrow or wide ranges can be used to meet the project or contract needs. MDLs are easily 10-fold lower than the MDLs cited in the method, and a high level of precision and accuracy can be expected across any calibration range, particularly at the lower concentrations. For full study results, ask for a copy of Reference #1.

## Reference

1. Shimadzu Guide to US EPA Method 624 for Analysis of Volatile Organic Compounds in Wastewater, GCMS Application News No. GCMS-1406.
2. Shimadzu Guide to BFB Tuning for Analysis of Volatile Organic Compounds, GCMS Application News No. GCMS-1405.
3. Appendix A to Part 136, Methods for Organic Chemical Analysis of Municipal and Industrial Wastewater, Method 624 – Purgeables.
4. Definition and Procedure for the Determination of the Method Detection Limit. *Fed. Regist.* **1984**. 49 (209), Appendix B to Part 136.